## Effects of experimental CO<sub>2</sub> leakage on solubility and transport of seven trace metals in seawater and sediment.

Murat V. Ardelan, Eiliv Steinnes, Syverin Lierhagen and Sven Ove Linde

## **Supporting information**

Figure S1 Figure S2 Figure S3 Figure S4 Figure S5 Figure S6 Figure S7 Figure S8

## pH, TIC, Tot. Alk and pCO2

pH and total alkalinity (Talk) were measured regularly with a combined glass electrode with an Ag/AgCl reference electrode (Radiometer), and partial pressure /fugacity of CO<sub>2</sub> (pCO<sub>2</sub>), and total inorganic carbon (TIC) were calculated based on the measurements of pH and total alkalinity (Öztürk et al, 2003). The pH electrode was calibrated using NBS buffer solutions at 6 °C. In order to convert measured pH (pH<sub>NBS</sub>) to pH<sub>total</sub>, the apparent activity coefficient of H<sup>+</sup> (f<sub>H+</sub>) was determined by four-point titration of 50 ml seawater by 15 ml standard acid with normality in the range of 0.008–0.014N HCl. Partial pressure /fugacity of CO<sub>2</sub> (P<sub>CO2</sub>), TIC was calculated after conversion of pH<sub>NBS</sub> to pH<sub>total</sub> (Öztürk et al, 2003).

## References

1. Öztürk, M., Vadstein, O., and Sakshaug, E. 2003. The effects of enhanced phytoplankton production on iron speciation and removal in mesocosm phases in a landlocked basin of Hopavågen, Norway. Marine Chemistry 84:3-17.



**Figure S1.:** Variation of pH (a and b), total alkalinity (c and d), Total inorganic carbon (e and f) and partial pressure/fugacity of  $CO_2$  (pCO<sub>2</sub>) (g and h) during both phases in CO2 seepage (full triangles) and in the control (open circles) chambers



**Figure S2:** Concentrations of total Al (Tot.Al) and dissolved Al (D.Al) in water during the first (a, b) and second (c, d) phases in CO<sub>2</sub> seepage (full triangles) and in the control (open circles) chambers.



**Figure S3:** Concentrations of total Cr (Tot. Cr) and dissolved Cr (D.Cr) in water during the first (a, b) and second (c, d) phases in CO<sub>2</sub> seepage(full triangles) and in the control (open circles) chambers.



**Figure S4:** Concentrations of total Ni (Tot. Ni), and dissolved Ni (DNi) in water during the first (a, b) and second (c, d) phases in CO<sub>2</sub> seepage(full triangles) and in the control (open circles) chambers.



**Figure S5:** Concentrations of total Cd (Tot.Cd) and dissolved Cd (DCd) in water during the first (a, b) and second (c, d) phases in CO<sub>2</sub> seepage(full triangles) and in the control (open circles) chambers.



**Figure S6:** Concentrations of total Zn (Tot.Zn) and dissolved Zn(D.Zn) in water during the first (a, b) and second (c, d) phases in CO<sub>2</sub> seepage(full triangles) and in the control (open circles) chambers.



**Figure S7:** Concentrations of total Cu (Tot.Cu) and dissolved Cu (D.Cu) in water during the first (a, b) and second (c, d) phases in CO<sub>2</sub> seepage(full triangles) and in the control (open circles) chambers.



**Figure S8:** Concentrations of total Pb (Tot.Pb) and dissolved Pb (D.Pb) in water during the first (a, b) and second (c, d) phases in CO<sub>2</sub> seepage(full triangles) and in the control (open circles) chambers.